

Name: \_\_\_\_\_

Date: \_\_\_\_\_

## Science 10 – Writing Ionic Formulas

### Goal

I can write ionic formulas for simple binary ionic compounds.

### Part A

1. Write the Lewis diagrams for \_\_\_\_\_ and \_\_\_\_\_.
2. When the metal becomes an ion, what happens?
3. When the non-metal becomes an ion, what happens?
4. How does this happen in this situation?
5. What happens after both the metal and non-metal become ions?

### Part B

1. Use the ion tiles to model the compound for Part A. Draw a picture.

2. Fill in this chart for the compound in Part A.

Name	Cation	Anion	# Cations	# Anions	Formula

Name: \_\_\_\_\_

Date: \_\_\_\_\_

**Part C**

Build each of the following compounds using the ion tiles, then fill in the chart for each.

<b>Name</b>	<b>Cation</b>	<b>Anion</b>	<b># Cations</b>	<b># Anions</b>	<b>Picture of Tiles</b>	<b>Formula</b>
magnesium chloride						
sodium oxide						
magnesium oxide						
aluminum chloride						
sodium phosphide						
aluminum oxide						
magnesium phosphide						
aluminum phosphide						