

Bohr-Rutherford Diagrams

	Group 1												Group 18					
Period 1																		
Period 2																		
Period 3																		
Period 4																		

Questions

For each question, consider only the electrons in the outermost (valence) shell.

How does the number of electrons change across a period (row)?

What is the relationship between the group number and the number of electrons?

What is the relationship between the period number and the number of shells?

